THE THREE-PHASE TEST: THE CONANT-SWAN REACTION

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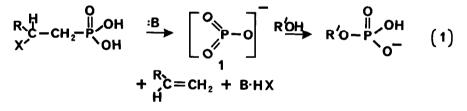
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The phosphorylations observed during the base-induced decomposition of β -halo phosphonates (the Conant-Swan reaction¹) may proceed by direct nucleophilic attack on the phosphorous or <u>via</u> the intermediacy of the monomeric metaphosphate ion, 1. While reasonable arguments² and experimental data³ are in accord with the indicated mechanism (eq. 1), detection of an intermediate would strengthen the postulate that these reactions generate 1. Reactive intermediates

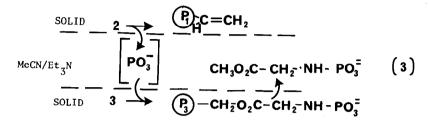


in other phosphate transfer reactions have been exposed using three-phase methods 4 and here we report application of this technique to the Conant-Swan reaction.

The polymer-bound phosphonate 2, (eq. 2) was prepared by successive treatment of chloromethylated, 2% crosslinked polystyrene (Merrifield's resin) with Ph_3P and $H_2CO/NaOH^5$, followed by PCl_5 and then hydrolysis^{6(1d)}. Merrifield resin beads of a different mesh were used to prepare the glycine derivative 3 following the established methods of solid phase peptide synthesis.

$$(\mathbf{P}_{2}^{-}-\mathbf{CH}_{2}^{-}-\mathbf{CI}) \xrightarrow{\mathbf{P}_{1}}_{\mathbf{H}} \mathbf{C} = \mathbf{CH}_{2} \xrightarrow{\mathbf{P}_{2}}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{CH}_{2}^{-}-\mathbf{P}_{2}^{-})}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-})}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-})}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-})}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-})}_{\mathbf{CI}} \underbrace{(\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-}-\mathbf{P}_{2}^{-$$

Suspension of the two solid phases in MeCN containing triethylamine resulted in phosphorylation (56% phosphate transfer in 4 hrs) of the polymer-bound amine. Separation (by screening) and transesterification $(MeOH/Me_2N-(CH_2)_2OH)$ liberated the methyl ester of glycine N-monophosphate, identical with an authentic sample (eq. 3).



The monophosphorylation of 3 requires an intermediate in solution since direct reaction of the two solid phases does not occur. The present experiment does not permit the assignment of a unique structure to the actual phosphorylating agent even though the intermediate exhibits the behavior expected for monomeric metaphosphate.⁷ The recent isolation⁸ and characterization⁹ of a metaphosphate derivative underscores the importance of these species in phosphate transfer reactions.

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